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Fast, ultra-trace detection of juvenile hormone III from mosquitoes using mass spectrometry

Cesar E. Ramirez

Department of Chemistry and Biochemistry, Florida International University, cramirez@fiu.edu

Marcela Nouzova

Department of Biological Sciences and Biomolecular Sciences Institute, Florida International University, nouzovam@fiu.edu

Paolo Benigni

Department of Chemistry and Biochemistry, Florida International University, pbenigni@fiu.edu

Martin Quirke

Department of Chemistry and Biochemistry, Florida International University, quirke@fiu.edu

Fernando G. Noriega

Department of Biological Sciences and Biomolecular Sciences Institute, Florida International University, noriegaf@fiu.edu

See next page for additional authors

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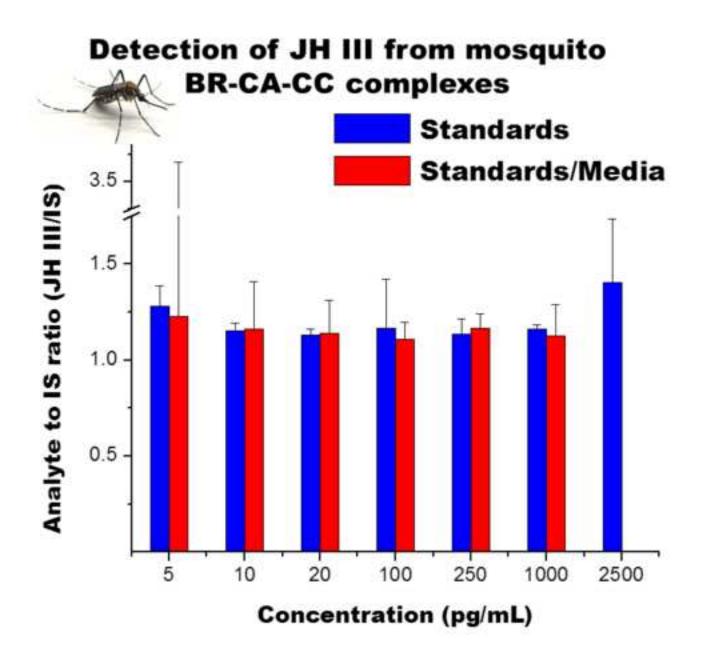
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Authors Cesar E. Ramirez, Marcela Nouzova, Paolo Benigni, Martin Quirke, Fernando G. Noriega, and Francisco Fernandez-Lima

*Highlights (for review)

Highlights

- A new protocol for fast separation and quantification of JH III from biological samples using liquid chromatography coupled to electrospray tandem mass spectrometry is described.
- the proposed protocol improves existing methodologies by combining a limited number of sample preparation steps with fast LC-MS/MS detection, providing a 8 pg/mL (0.32 pg on column) limit of detection (15-fold gain in sensitivit) with high inter and intraday reproducibility.
- A detailed description of the JH III fragmentation pathway is provided for the first time, based on isolation of the molecular ion and their intermediate fragments using in-source MS/MS, MS/MSⁿ and FT-ICR MS/MS measurements.
- The performance of the LC-MS/MS protocol is comparable to previously described JH III quantitation protocol based on fluorescence detection, with the added advantage that quantification is independent of the availability of fluorescent tags that are often unavailable or show quite diverse responses on a batch-to-batch basis.
- The JH III workflow was evaluated as a function of developmental changes, sugar feeding and farnesoic acid stimulation in mosquitoes.



- 1 Fast, ultra-trace detection of Juvenile Hormone III from mosquitoes using
- 2 mass spectrometry
- 3 Cesar E. Ramirez¹, Marcela Nouzova^{2,3}, Paolo Benigni¹, J. Martin E. Quirke¹, Fernando G.
- 4 Noriega^{2,3}, and Francisco Fernandez-Lima^{1,3}*
- 5 ¹Department of Chemistry and Biochemistry, Florida International University, Miami, USA
- 6 ²Department of Biology, Florida International University, Miami, USA
- ³Biomolecular Science Institute, Florida International University, Miami, USA
- **8 Corresponding Author**
- 9 *Phone: 305-348-2037. Fax: 305-348-3772. E-mail: fernandf@fiu.edu.
- 10 Notes
- 11 The authors declare no competing financial interest.
- 12 ABSTRACT
- 13 In the present work, a new protocol for fast separation and quantification of JH III from
- biological samples using liquid chromatography coupled to electrospray tandem mass
- spectrometry is described. In particular, the proposed protocol improves existing methodologies
- by combining a limited number of sample preparation steps with fast LC-MS/MS detection,
- providing lower limits of detection and demonstrated matrix effect control, together with high
- inter and intraday reproducibility. A limit of detection of 8 pg/mL (0.32 pg on column) was
- achieved, representing a 15-fold gain in sensitivity with respect to previous LC-MS based
- protocols. The performance of the LC-MS/MS protocol is comparable to previously described
- 21 JH III quantitation protocol based on fluorescence detection, with the added advantage that
- 22 quantification is independent of the availability of fluorescent tags that are often unavailable or
- show quite diverse responses on a batch-to-batch basis. Additionally, a detailed description of
- the JH III fragmentation pathway is provided for the first time, based on isolation of the

 molecular ion and their intermediate fragments using in-source MS/MS, MS/MSⁿ and FT-ICR MS/MS measurements. The JH III workflow was evaluated as a function of developmental changes, sugar feeding and farnesoic acid stimulation in mosquitoes and can be applied to the detection of other juvenile hormones.

Keywords: Juvenile hormone III, Liquid chromatography, MRM, quantification, extraction.

INTRODUCTION

Juvenile hormones (JHs) are synthesized by the *corpora allata* (CA) and play key roles in many processes in insect development and reproduction, including inhibition of metamorphosis, caste determination and differentiation, stimulation of flight and migration, stimulation of reproduction, regulation of diapause, stress resistance, and aging[1-6]. Several JHs have been identified and characterized in insects, with JH III being the most widespread[7-9, 6]. A common structural feature for all JHs is the presence of an epoxide group near one end and a methyl ester on the other. JH titers in small insects are often in the femtomole to picomole range, which makes it challenging to detect by most typical analytical techniques[10-14].

The most widely used analytical methods for identification and detection of JHs include nuclear magnetic resonance (NMR), infrared spectroscopy (IR), and gas chromatography (GC)—mass spectrometry (MS) [14-19]. During GC-MS, fragment ions from the electron or chemical ionization process are typically used to identify the JH molecules; however, this analysis typically requires lengthy preparation steps[20, 19, 21-23]. More recently, several studies have shown the advantages of a number of additional techniques for the identification and quantification of JHs, such as direct analysis in real time-MS (DART-MS), high performance liquid chromatography-MS/MS (HPLC- MS/MS), HPLC with fluorescence detection (HPLC-FD) and ultra-performance liquid chromatography-MS (UPLC-MS)[18, 24-32]. A variety of ionization sources have been utilized prior to the MS analysis, such as electrospray ionization (ESI), atmospheric pressure chemical ionization (APCI), atmospheric pressure photoionization (APPI) and atmospheric pressure thermospray ionization (APTSI); allowing the detection of the JH III molecular ion in the protonated and sodiated forms[33]. While some studies showed the

 advantages of using the sodiated species for quantification in single ion monitoring mode (SIM), only the protonated species allowed for multiple reaction monitoring (MRM)[25].

In the present work, we describe an extraction protocol followed by an HPLC-ESI-MS/MS analysis that increased sensitivity and reproducibility, while reducing the analysis time for JH III detection in mosquito samples. The features of the method proposed are compared with previously established MS- and FD-based methods. Additionally, a detailed description of the fragmentation behavior of JH III [M+H]⁺ molecular ion is described for the first time.

MATERIALS AND METHODS

Materials and reagents

Certified standard solutions for JH III and its deuterated analog (JH III-D3) were obtained from Toronto Research Chemicals (Toronto, Canada). Sodium chloride, potassium chloride, hydrochloric acid, sodium hydroxide, ammonium acetate, ammonium formate and ammonium hydroxide salts were analytical grade or better (Fisher Scientific, Pittsburgh, PA). Water, methanol, hexane and acetonitrile were all Optima grade or better (Fisher Scientific). Chromatographic mobile phases (0.1% formic acid in water, and 0.1% formic acid in acetonitrile) of Optima LC-MS grade were also purchased from Fisher Scientific, and used as received. Tissue culture media Gibco M-199, silanized LC vials and silanized LC vials with fused 250 µL inserts were also purchased from Fisher Scientific. The tuning mix calibration standard (G24221A) was obtained from Agilent Technologies (Santa Clara, CA).

Sample preparation and storage

Biological samples were prepared following the protocol described in Figure 1. Briefly, preparations were of intact *corpora allata-corpora cardiaca* (CA-CC) complexes connected to the brain and head capsule, and are denoted as BR-CA-CC complexes. BR-CA-CC were dissected in a drop of mosquito saline-buffer containing 138 mM NaCl, 8.4 mM KCl, 4 mM CaCl₂, 12 mM NaH₂PO₄ and 42.5 mM sucrose[34]. After dissection, the BR-CA-CC complexes were incubated in 150 μL of tissue culture media M-199, containing 2% Ficoll 400 and 50 μM methionine. Incubations of BR-CA-CC complexes were carried out in a humid chamber in silanized 2 mL vials for 4 h in the dark at 32°C, and under continuous gentle agitation. After

incubation, 10 µL of 6.25 ppb JH III-D3 in acetonitrile were added to each sample, followed by 600 µL of Hexane. Samples were vortexed for 1 minute, and spun for 5 minutes at 4°C and 2000 g. The organic phase was transferred to a new silanized vial and dried under nitrogen flow. Dried extracts were re-suspended in 100 µl of acetonitrile, vortexed 1 minute, transferred to a new silanized vial with a fused 250 µL insert and stored at -20°C.

Mass Spectrometry Analysis

The mass spectrometry analyses were carried out in-house using HPLC-MS/MS. Briefly, sample injections (40 µL) and LC separations were performed by a Prominence LC-20AD Ultra-Fast Liquid Chromatograph (Shimadzu, Kyoto, Japan), equipped with a Dionex Acclaim 120 C18 Column (250x2.1 mm, 5 µm) obtained from Thermo Scientific (Sunnyvale, CA). Column temperature was kept at 40°C. A 15 minutes binary gradient program between 0.1% formic acid dissolved in water (mobile phase A) and 0.1% formic acid dissolved in acetonitrile (mobile phase B) was run according to the following timetable: hold 5% B for 0.5 min; ramp to 98% B in 7.5 min; hold 98% B for 3 min; return to 5% B in 0.5 min; hold 5% B for 3.5 min. Flow rate was constant at 0.8 mL/min. Detection was performed by a QTRAP® 5500 triple quadrupole mass spectrometer (AB Sciex, Ontario, Canada) equipped with a Turbo VTM ion source. The mass spectrometer was operated under ESI positive mode ionization with multiple reaction monitoring (MRM) of two transitions per compound. MRM detection and electrospray source parameters were optimized by infusing 1 mg/L solutions of each compound in 0.1% formic acid in acetonitrile. Source parameters were: curtain gas = 10 psi; spray Voltage = 5000 V; temperature 400°C; ion source gas 1= 40 psi; ion source gas 2= 50 psi; entrance potential = 7.0 V. MRM transitions parameters are listed in Table 1.

JH III and JH III-D3 chemical structures and purities were confirmed by MS/MS and accurate mass measurements using Fourier transform ion cyclotron resonance-MS (nanoESI-FT-ICR MS). In-source fragmentation and MS/MSⁿ capabilities in the OTRAP® 5500 triple quadrupole mass spectrometer (AB Sciex, Ontario, Canada) were used to determine the fragmentation pathways of JH III and JH III-D3 (see Table 1 and 2). nanoESI FT-ICR MS/MS experiments were performed in positive ion mode in a 7T Solarix FT-ICR MS spectrometer (Bruker Daltonics, Inc., Billerica, MA). Ion transmission was optimized for high sensitivity for JH III and its fragments in the m/z 100–300 range. Ions were accumulated in the collision cell (2

125 35

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51 52 **135**

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MHz, 1000 Vpp) for 4s. FT-ICR MS spectra were acquired over 200-time domain acquisitions at 2 Mword (1 s transient). FT-ICR signals were processed using a half-sine apodization followed by Fast-Fourier transform and broadband phase correction (absorption spectra using absorption mode processing, AMP), resulting in a 2-fold increase in mass resolution.

RESULTS AND DISCUSSION

The main aim of this study was to develop a protocol for fast and accurate measurement of JH III from biological samples using mass spectrometry. In particular, our efforts were focused on: 1) minimizing the number of steps during sample preparation prior to the MS analysis, 2) developing a fast LC-MS/MS protocol, 3) improving sample storage protocols, 4) evaluate inter and intraday analysis variation, 5) refining the interpretation of the JH III MS/MS fragmentation pathway, 6) enhancing limits of detection while reducing matrix effects, and 7) comparing the performance with a previously utilized JH III quantitation protocol based on fluorescence detection[30].

Mass Spectrometry Analysis

JH III can be typically detected in the protonated and sodiated forms (e.g., m/z = 267 [M+H]⁺ and m/z = 289 [M+Na]⁺), depending on the spraying and solvent conditions[24]. Since the fragmentation of the sodiated molecular ion [M+Na]⁺ does not produce diagnostic fragment ions, previous studies have preferred LC-MS for JH identification and quantitation, with the caveat that quantification accuracy may be compromised by potential interferences due to the complex nature of the biological sample[28, 29]. Alternatively, the fragmentation of the protonated form m/z = 267 [M+H]⁺ provides a variety of signature fragmentation pathways that can be used to identify the JH III structure (Figure 2 top). The [M+H]⁺ molecular ion can undergo fragmentation via collision induced activation using in-source MS/MS and the MS/MSⁿ linear trap region of the LC-QQQ (see Table 2). Inspection of the in-source and MS/MSⁿ data permitted the construction of the JH III [M+H]⁺ fragmentation pathway (Figure 3). Upon collision induced activation, the JH III [M+H]⁺ main fragmentation pathways lead to the observation of m/z = 235 [M+H-CH₃OH]⁺ and m/z = 249 [M+H-H₂O]⁺ ions, corresponding to the loss of one methanol group and one water molecule, respectively. Further activation of m/z = 235 and 249 generates m/z = 217 [M+H-CH₃OH-H₂O]⁺ and m/z = 207 [M+H-CH₃OH-CO]⁺,

 corresponding to the combined loss of one methanol group and water, and to the loss of a CO group, respectively. In contrast to previous reports [25], our data support the generation of m/z =217 as a product of 267->235->217 and 267->249->217. The activation of m/z=217 produces $m/z = 189 \text{ [M+H-CH}_3\text{OH-H}_2\text{O-CO]}^+$, which further fragments into $m/z = 147 \text{ [M+H-CH}_3\text{OH-H}_2\text{O-CO]}^+$ H₂O-CO-CH₃CH=CH₂]⁺. Complementary analysis using nESI-FT-ICR MS/MS supports the above proposed mechanism. The nESI-FT-ICR MS/MS showed the fragmentation of [M+H]⁺ $(m/z=267.195461, C_{16}H_{27}O_3)$ into $[M+H-H_2O]^+$ $(m/z=249.18492, C_{16}H_{25}O_2), [M+H-CH_3OH]^+$ $(m/z=235.16929, C_{16}H_{23}O_2), [M+H-CH_3OH-H_2O]^+ (m/z=217.158786, C_{15}H_{21}O), [M+H-CH_2O]^+ (m/z=217.15876, C_{15}H_{21}O), [M+$ $CH_3OH-CO]^+$ (m/z = 207.174417, $C_{14}H_{23}O$), $[M+H-CH_3OH-H_2O-CO]^+$ (m/z = 189.163858, $C_{14}H_{21}$) and [M+H-CH₃OH-H₂O-CO-CH₃CH=CH₂]⁺ (m/z= 147.116827, $C_{11}H_{15}$) with sub-ppm mass accuracy (see Table 2). Previous studies utilizing chemical ionization ion trap mass spectrometry, performed the quantification based on the summed intensities of six diagnostic ions (m/z=235, 217, 189, 147, 125, and 111)[19]. Although this approach is also feasible in the case of LC-QQQ instruments, using a lower number of MRM transitions allows the monitoring of a higher number of points across the LC peak relative to the aforementioned MRM strategies and therefore provides higher sensitivity. The selection of the transitions to perform the JH III quantification was based on their relative abundance. That is, the most abundant fragmentation transitions utilized were 267->235 (primary) and 267-> 147 (secondary). Previous reports have also utilized 267-> 235 as a primary and 267-> 217 as secondary MRM transitions for LC-MS/MS analysis[25, 27].

The heavy isotopomer JH III-D3 was utilized as an internal standard to normalize all sample preparation, extraction and analysis steps in order to accurately quantify the amount of JH III hormone produced by the BR-CA-CC complexes. The JH III-D3 molecular ion was observed in the protonated form $[D+H]^+$, and the stoichiometry was confirmed utilizing nESI-FT-ICR MS (m/z = 270.21420) with sub-ppm accuracy. The fragmentation of the JH III-D3 $[D+H]^+$ molecular ion showed multiple similarities to JH III $[M+H]^+$ (Figure 2). That is, the loss of the deuterated methanol group led to the observation of m/z = 235 $[M+H-CD_3OH]^+$; further activation showed the same fragmentation pattern as the JH III $[M+H]^+$ molecular ion.

A plausible mechanistic sequence for the generation of JH III fragment ions is shown in Figure 3 and in the supplementary materials. Briefly, it is proposed that opening of the oxirane to form a tertiary alcohol and an alkene would facilitate the elimination of water (perhaps, via a 6-

electron cyclic transition state) leading to the formation of $m/z = 249 \, [M+H-H_2O]^+$ from m/z = $[M+H]^+$ and $m/z = 217 [M+H-CH₃OH-CO]^+$ from $m/z = 235 [M+H-CH₃OH]^+$. Similarly, elimination of methanol accounts for the formation of both $m/z = 235 \, [M+H-CH_3OH]^+$ from m/z= 267 $[M+H]^+$ and $m/z = 217 [M+H-CH_3OH-CO]^+$ from $m/z = 249 [M+H-H_2O]^+$. The loss of methanol is confirmed by the aforementioned isotopic labeling study with JH III-D3. In each case, the exclusion of the methanol generates a molecular ion in which the positive charge will reside on the C=O oxygen. The subsequent loss of CO from $m/z = 217 \text{ [M+H-CH}_3\text{OH-CO]}^+$ to form m/z = 189 observed by the nESI-FT-ICR MS/MS analysis is unlikely to occur on the basis of direct cleavage from an sp^2 carbon. Thus, it is proposed that a rearrangement of the C-2 double bond occurs, allowing the CO to leave with the generation of a resonance-stabilized allylic carbocation that has the potential to undergo a Wagner-Meerwein hydride shift to form an even more stable allylic carbocation in which the positive charge is delocalized over an entire 7carbon conjugated triene array. Whether the rearrangement and elimination are separate steps or a concerted process is not clear. Further rearrangement of this system allows the elimination of propene as a neutral fragment generating m/z = 147 [M+H-CH₃OH-H₂O-CO-CH₃CH=CH₂]⁺ asan even more delocalized allylic carbocation in which the positive charge can be delocalized over the entire 9-carbon tetraene unit. The loss of CO from $m/z = 235 \, [M+H-CH_3OH]^+$ to generate $m/z = 207 \text{ [M+H-CH}_3\text{OH-CO]}^+$ is proposed to occur via a similar initial rearrangement of the C-2 double bond. The resultant allylic carbocation will also be resonance stabilized. It is unclear whether the loss of CO occurs on the oxirane or the rearranged oxirane or both.

HPLC-MS/MS method validation

The HPLC-MS/MS workflow was developed using standards and biological samples, with the extraction protocol described in Figure 1. The HPLC consisted of a short 15 minutes run with JH III eluting at around 8.2 minutes (Figure 4). As expected, JH III and its heavy analog JH III-D3 co-eluted, and their detection was based on monitoring the 267->235 and 270->235 as primary and 267->147 and 270->147 as secondary transitions. Notice that the primary MS/MS channel is 4-5 fold more abundant that the secondary channel. Comparison between the standard and the biological samples showed no changes on retention time, as well as on the ratio between transitions, allowing the use of these parameters for analyte confirmation purposes.

The effect of analyte extraction was evaluated by comparing the instrument response of analyte to internal standard ratio (JH III/JH III-D3), for pure- and media-extracted standards over a wide concentration range (Figure 5a). An extraction recovery of near 55% was routinely observed regardless of the analyte concentration. No improvements were observed by performing a double extraction with hexane (Figure 5b). The reduced sensitivity caused by the extraction efficiency is evidenced in the reduced slope on the linear response of the instrument as a function of analyte concentration (Figure 5c), justifying the use of JH III-D3 to normalize this step.

The HPLC-MS/MS method limit of detection (LOD), defined as 3 times the background analytical response, was determined experimentally to be 8 pg/mL (0.32 pg on column) by spiking increasing quantities of JH III in ACN (at 5, 10, 20, 100, 250 and 1000 pg/mL in triplicates, n=18), and calculating the uncertainty on the intercept of the obtained curve. A LOD of 19 pg/mL (0.76 pg on column) was obtained by spiking the analyte into M-199 medium and submitting the prepared solutions to the complete extraction protocol; this higher LOD is a consequence of a reduced signal after extraction. However, the obtained protocol LOD is approximately 10 times lower than previous LC-MS based methodologies for the detection of JH III in hemolymph (LOD = of 4-5 pg on column[29, 27, 31]).

Inter and intra-day reproducibility was evaluated by extracting and analyzing spiked M-199 media with 500 pg/mL of JH III. The protocol was halted at the drying step (Figure 1) and the extracted standards were stored dry for 0 (reference, n=32), 1, 4, 5 and 7 days (n=5 each). There were no significant differences between the measurements in terms of peak areas or recovery (Figure 6a), which suggested that the analyte extracted from spiked media could be stored dry for at least 7 days. Most intra-day variability occurred between one standard deviation from the pooled mean of all measurements. A relative standard deviation (RSD) of 10% or lower was obtained for each analysis day, while an RSD of 16% was obtained in the complete dataset.

Analysis of JH III from biological samples

In order to evaluate the applicability of this analysis protocol, JH III synthesized in vitro by mosquito BR-CA-CC complexes under different developmental and physiological conditions were analyzed. Samples were prepared following the protocol previously described (Figure 1). Calibration curves were obtained by plotting the peak area ratio of JH III to JH III-D3 as a

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⁴⁸ **254**

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⁵⁹ **260**

 function of the analyte concentration. Linearity was observed over a wide concentration range $(R^2 > 0.999; 5 \text{ to } 2500 \text{ pg/mL}, 0.2 \text{ pg to } 100 \text{ pg on column})$. The concentration of JH III-D3 was constant in biological samples and calibration solutions (625 pg/mL). Calibration standards were analyzed in duplicate at the beginning and at the end of each analytical sequence, with calibration verification performed every 12 runs by injecting 250 pg/mL of JH III in acetonitrile buffer, allowing a 15% deviation. A control blank (non-extracted) was analyzed with every batch, consistently displaying the absence of JH III signal. The analysis sequence also included carryover tests by verifying the absence of a JH III signal upon injecting the control blank after the highest calibration standard, as well as after every calibration verification standard.

In each LC-QQQ analysis batch, duplicates of a negative control and two concentrations of positive controls (50 pg/mL and 500 pg/mL) were extracted according to Figure 1 and measured, with deviations of 15% or lower consistently observed. JH III compound identifications were considered positive when a signal-to-noise ratio larger than 3 were present in both the quantification and confirmation MRM transitions, with a difference of less than 0.05 min in retention time relative to that of the JH III-D3. The results of the analysis of over 1500 samples revealed that the confirmation ratios for internal standards and biological samples were constantly within two standard deviations (Figure 5d).

In order to investigate the effect of sample storage on analyte signal, 10 biological samples were split after hexane extraction and the resulting extract aliquots were dried under nitrogen. One of the dried aliquots was stored at 4°C and the second was reconstituted in acetonitrile for immediate analysis. Five of the stored aliquots were reconstituted and analyzed after 5 days, while the remaining samples were analyzed after 30 days of storage. There were no significant differences in terms of recovery relative to the initial measured concentration (Figure 6b) as a function of storage time, suggesting that dried biological samples can be correctly quantified after a month of dry storage. Previous work has shown that JH III is stable for up to one month and up to six 6 months in methanol if stored at 4°C and -18°C, respectively[25]. Similar stabilities were detected when JH III was stored in hexane and acetonitrile. However, the possibility to store dry samples largely facilitates the handling and shipping for inter-laboratory comparisons and complementary measurements.

Further validation of the JH III quantitation protocol was attained by analyzing JH III synthesis by BR-CA-CC adult female mosquito preparations in vitro as a function of

 developmental changes, sugar feeding and farnesoic acid stimulation. The results of these experiments were compared with previous studies using HPLC-FD[35-37]. The comparison of the MS and the FD results showed good agreement for all the cases, considered reinforcing the value of the new developed protocol to assess JH III levels in mosquitoes (Figure 7). The new LC-MS/MS protocol showed similar high sensitivity, accuracy and reproducibility compared to the previously reported HPLC-FD[30], with the advantage of having selective MS detection that enables absolute quantification by using a heavy analog as an internal standard. The HPLC-FD protocol, while also sensitive, depends on commercially made tags that are often unavailable, require derivatization steps and show quite diverse responses on a batch-to-batch basis. On the contrary, this LC-MS/MS protocol includes a straightforward extraction protocol followed by a sensitive and reproducible HPLC-ESI-MS/MS analysis for absolute JH III detection in mosquito samples.

CONCLUSIONS

An analytical workflow for fast, ultra-trace quantitation of JH III from biological samples was developed. The fragmentation pathway of JH III [M+H]+ molecular ion was studied, and a mechanistic model is proposed. A HPLC-MS/MS workflow based on MRM using 267-> 235 and 267 -> 147 was optimized for quantitative analysis, with higher sensitivity and a reduced number of sample preparation steps. A better analytical performance of the proposed protocol was demonstrated in terms of reproducibility, sensitivity and routine applicability, with higher sensitivity than previous LC-MS applications for the detection of JH III and with similar sensitivity to spectrofluorometric methods without the need of lengthy derivatization steps. Results showed that the storage protocol allows for quantification of samples with at least one month of dry storage. Further developments of the proposed workflow can be used to the analysis of other juvenile hormones.

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- 58 288 International University.

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⁴ 315

- pupae and adult female mosquitoes at different hours before and after adult eclosion. FD values
- are from reference[35]. B) Effect of sugar feeding. MS: Newly eclosed adult females were fed
- for 4 days on 3% or 20% sucrose. FD values are from reference[37]. C) MS: BR-CA-CC
- preparations were dissected from 4 day-old adult female mosquitoes and stimulated with 40μM
- farnesoic acid (FA). FD values are from reference[36].

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Table 1. Common MRM parameters utilized for JH III and JH III-D3 detection on the QTRAP® 5500 triple quadrupole mass spectrometer.

Compound	$DP^{1}(V)$	Q1 (Da)	$CE^{2}(V)$	Q3 (Da)	CXP3 (V)
JH III	46	267	11	235	30
			19	147	16
JH III-D3	46	270	11	235	28
			17	147	6

¹ Declustering Potential; ² Collision Energy; ³ Collision cell exit potential.

Table 2. Typical fragment ion and dissociation pathways observed using in-source MS/MS, MS/MSⁿ and FT-ICR MS/MS for JH III [M+H]⁺ molecular ion.

	Parent m/z	Fragment m/z	
	267	249, 235, 217,	
	207	189, 147	
In-source	249	217, 189, 147	
MS/MS	235	217, 189, 147	
	217	189, 147	
	189	147	
	267 > 225	217 207 100	
	267 -> 235	217, 207, 189	
MS/MS ⁿ	267 -> 249	217, 189, 147	
	267 -> 217	189, 147	
	267 -> 189	147	
	267.195461	249.18492	
	$C_{16}H_{27}O_3$	$C_{16}H_{25}O_2$	
		235.169298	
		$C_{15}H_{23}O_2$	
		217.158786	
FT-ICR MS/MS		$C_{15}H_{21}O$	
F1-ICK MS/MS		207.174417	
		$C_{14}H_{23}O$	
		189.163858	
		$C_{14}H_{21}$	
		147.116827	
		$C_{11}H_{15}$	

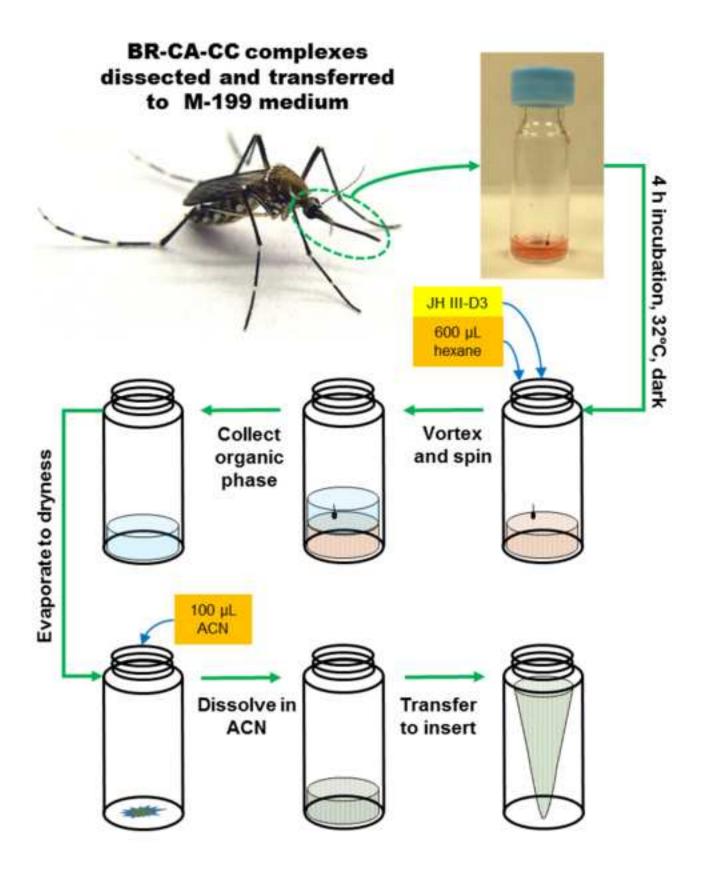


Figure 2
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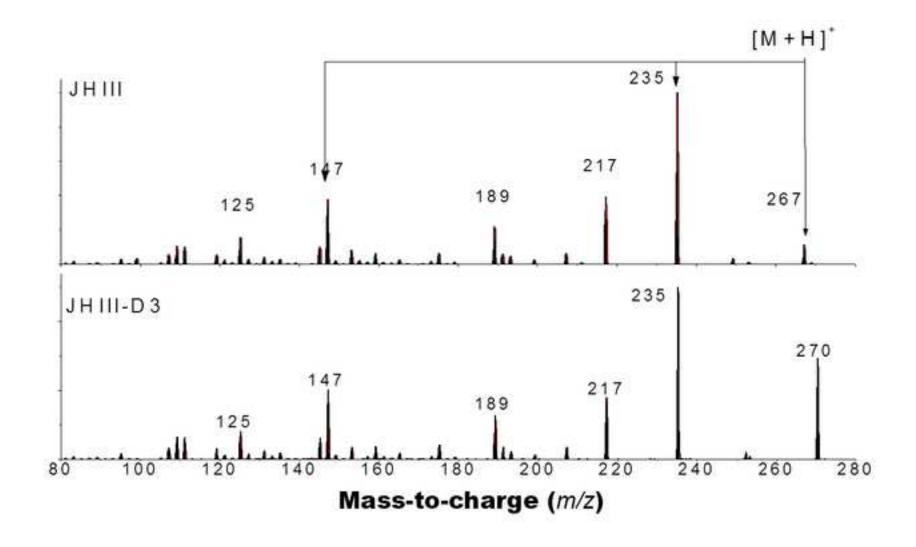


Figure 3
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Figure 4
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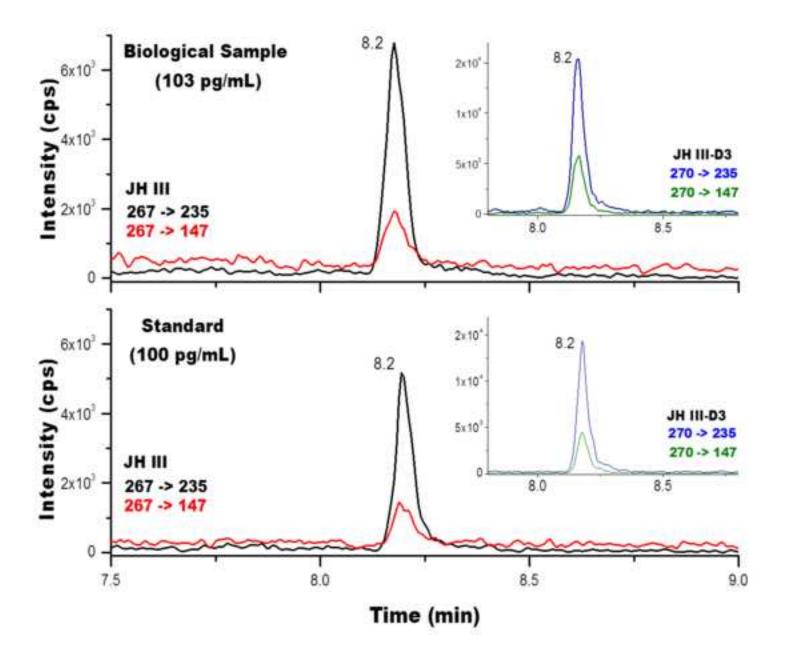
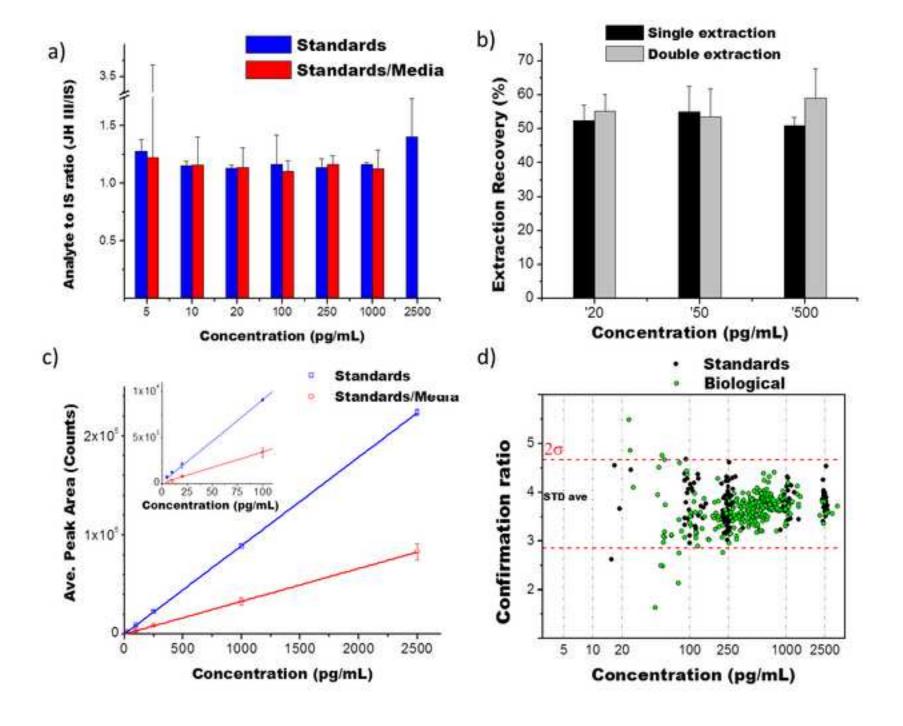
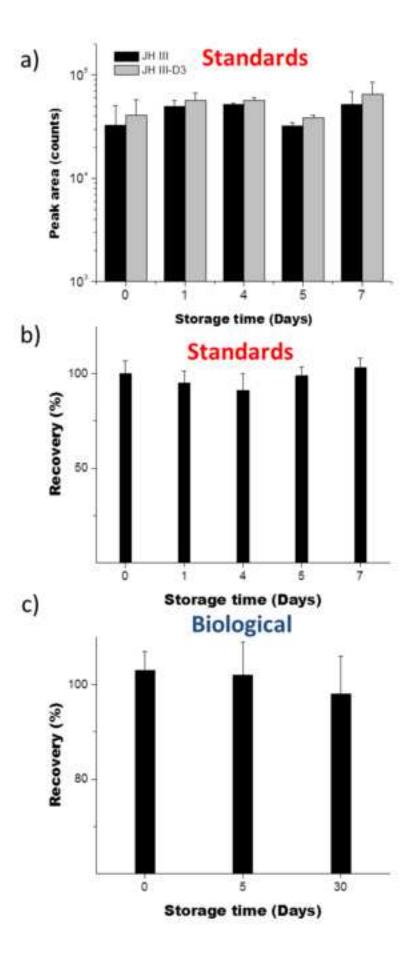
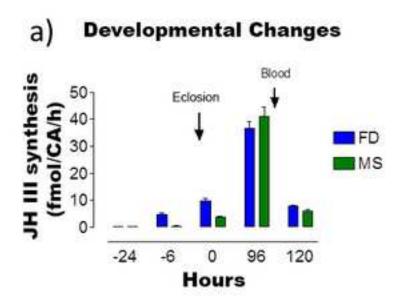
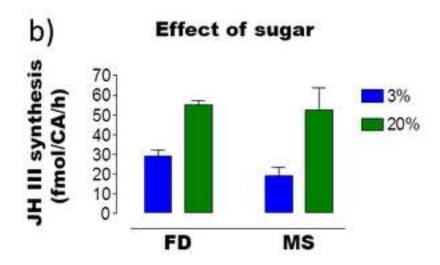


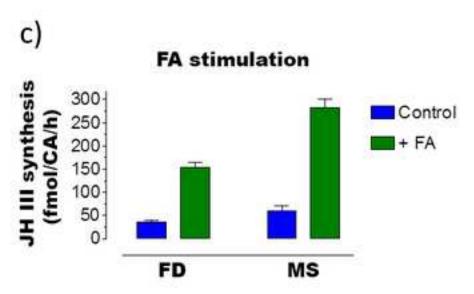
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